

Zumdahl 8th Edition Solutions

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Water has a very high specific heat capacity of 4.1814 J/(g·K) at 25 °C - the second highest among all the heteroatomic species (after ammonia), as well as a high heat of vaporization (40.65 kJ/mol or 2257 kJ/kg at the normal boiling point), both of which are a result of the extensive hydrogen bonding between its molecules. These two unusual properties allow water to moderate Earth's ...

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