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174 Study Guide for An Introduction to Chemistry a model (and therefore a simplification of reality), it is extremely useful. You will find the information in Table 12.1: Covalent Bonding Patterns very helpful when drawing Lewis structures (the task described in Section 12.2).

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Stoichiometry Study Guide The following example problems exhibit the types of calculations you will be expected to perform on tomorrow's test. •

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Chapter 12 - Chemical Kinetics . 12.1 Reaction Rates . A. Chemical kinetics 1. Study of the speed with which reactants are converted to products B. Reaction Rate 1. The change in concentration of a reactant or product per unit of time [] t A t t concentration of A at time t concentration of A at time t Rate $\Delta \Delta = - - = 2 1 2 1$. a ...

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The amount of product that actually forms when the reaction is carried out in the laboratory.

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Study Guide for Content Mastery 11. State the octet rule. ShLM 12. What does the electronegativity of an element indicate? 13. What are the period and group trends in electronegativities? 36

Chemistry: Matter and Change Chapter 6

Chapter 12 - Chemical Kinetics

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chemistry. • Describe oxidation and reduction (redox) reactions. • Discuss the different forms of matter: elements, compounds, and mixtures. • Explain the difference between solutions, suspensions, and emulsions. • Explain pH and the pH scale.

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